



# Facile synthesis of $\text{Fe}_3\text{O}_4$ @mesoporous $\text{TiO}_2$ microspheres for selective enrichment of phosphopeptides for phosphoproteomics analysis

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## ABSTRACT

Protein phosphorylation is one of the most important post-translational modifications. Due to the dynamic nature and low stoichiometry of the protein phosphorylation, enrichment of phosphopeptides from proteolytic mixtures is necessary prior to their characterization by mass spectrometry. In this work, we synthesized  $\text{Fe}_3\text{O}_4$ @mesoporous  $\text{TiO}_2$  magnetic microspheres with core-shell structure and large surface area for selective enrichment of phosphopeptides. To demonstrate its ability for selective enrichment of phosphopeptides, we applied  $\text{Fe}_3\text{O}_4$ @mesoporous  $\text{TiO}_2$  magnetic microspheres to isolation and enrichment of the phosphopeptides from tryptic digestion of standard proteins and real samples, and then the enriched peptides were analyzed by matrix-assisted laser desorption/ionization mass spectrometry (MALDI-MS) or liquid chromatography coupled to electrospray ionization mass spectrometry (LC-ESI-MS). Due to that the as-made  $\text{Fe}_3\text{O}_4$ @mesoporous  $\text{TiO}_2$  microspheres have large surface area, good dispersivity and biocompatibility, they have been demonstrated as a powerful tool for phosphoproteomics research.

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## 1. Introduction

Post-translational modifications of proteins control many biological processes. The reversible phosphorylation of proteins is recognized as an essential post-translational modification regulating cell signaling and ultimately functions of biological systems. The significance of protein phosphorylation for a multitude of biological processes has resulted in major efforts to study protein phosphorylation for decades [1–3]. Unfortunately, due to the small quantity of phosphoproteins, the low stoichiometry of phosphorylation and the suppression effect by nonphosphorylated peptides, it remains a major challenge to detect them by mass spectrometry which is the most important and powerful tool for the analysis of protein phosphorylation [4–6].

A variety of approaches, including immunoprecipitation, chemical-modification strategies, strong cation exchange chromatography, immobilized metal affinity chromatography (IMAC) and metal oxide affinity chromatography (MOAC) have been used for this purpose [7–12]. Among these techniques, IMAC and MOAC are most simple and convenient and have been widely studied.

Titania has been investigated intensively in various areas including inorganic pigment, sunscreen, photocatalysis, energy storage, conversion, electrochromics, sensor fields and also proteomics [13–17]. Commendable specific interaction with phosphate group

of phosphopeptides leads to their excellent enrichment efficiency of phosphopeptides [18–22]. The effectiveness of titania in practical applications varied considerably with its composition, crystallinity, morphology and texture of the material and specific surface area and mesoporosity [23–26]. Large specific surface area offers lots of interaction sites and then leads to good enrichment efficiency. Therefore, the controlled synthesis of titania materials with large specific surface area is badly in demand. Besides, magnetic materials have drawn a lot of attention in recent years because of their facility of separation and good biocompatibility [27–29]. Therefore synthesis of a magnetic titania material with large specific surface area is necessary.

In this work, we report the synthesis of monodisperse  $\text{Fe}_3\text{O}_4$ @mesoporous  $\text{TiO}_2$  with high specific surface area and application of them for the enrichment of phosphopeptides. They were evaluated with tryptic digests of both standard protein and a mouse brain lysate for the phosphoproteomics analysis. All the experimental results demonstrated a good enrichment efficiency and selectivity of the method we report here.

## 2. Materials and methods

### 2.1. Materials and chemicals

TFA was purchased from Merck (Darmstadt, Germany). ACN was HPLC grade from Fisher Scientific (Fairlawn, NJ, USA). Bovine  $\beta$ -casein, bovine serum albumin (BSA), ovalbumin (from chicken

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egg), trypsin (from bovine pancreas, TPCK treated), ammonium bicarbonate ( $\text{NH}_4\text{HCO}_3$ ), dithiothreitol (DTT), iodoacetamide (IAA) 2, 5-dihydroxybenzoic acid (DHB) and phenylmethylsulfonyl fluoride (PMSF) were provided by Sigma-Aldrich Chemical (St. Louis, MO, USA). The water used in this study was purified using a Milli-Q water purification system from Millipore (Bedford, MA, USA). All other chemicals and reagents were of analytical grade and were purchased from Shanghai Chemical Reagent.

## 2.2. Characterization of $\text{Fe}_3\text{O}_4$ @mesoporous $\text{TiO}_2$

The morphologies of the as-synthesized  $\text{Fe}_3\text{O}_4$ @mesoporous  $\text{TiO}_2$  microspheres were studied under transmission electron microscopy (JEM-2100F) and scanning electron microscopy (XL30), respectively. Powder X-ray diffraction (XRD) patterns were recorded on a Bruker D4 X-ray diffractometer with Ni-filtered  $\text{Cu K}_\alpha$  radiation (40 kV, 40 mA). Nitrogen adsorption and desorption isotherms were measured using Micromeritics ASAP 2020. The samples were degassed in a vacuum at 200 °C for 8 h prior to measurement. The Brunauer–Emmett–Teller (BET) method was utilized to calculate the specific surface areas ( $S_{\text{BET}}$ ) using adsorption data in a relative pressure range from 0.18 to 0.35. By using the Barrett–Joyner–Halenda (BJH) model, the pore volumes and pore size distributions were derived from the desorption branches of the isotherms, and the total pore volumes ( $V_t$ ) were estimated from the adsorbed amount at a relative pressure  $P/P_0$  of 0.992.

## 2.3. Preparation of standard protein digests

Bovine  $\beta$ -casein, ovalbumin and Bovine serum albumin (BSA) were dissolved in 25 mM  $\text{NH}_4\text{HCO}_3$  buffer at pH 8.3 and treated with trypsin (2%, w/w) for 16 h at 37 °C, respectively. Before the digestion, ovalbumin and BSA were reduced with DTT (final concentration 10 mM, 37 °C, 1 h) and carboxamidomethylated with iodoacetamide in the dark (final concentration 20 mM, 37 °C, 30 min). Bovine  $\beta$ -casein was digested directly.

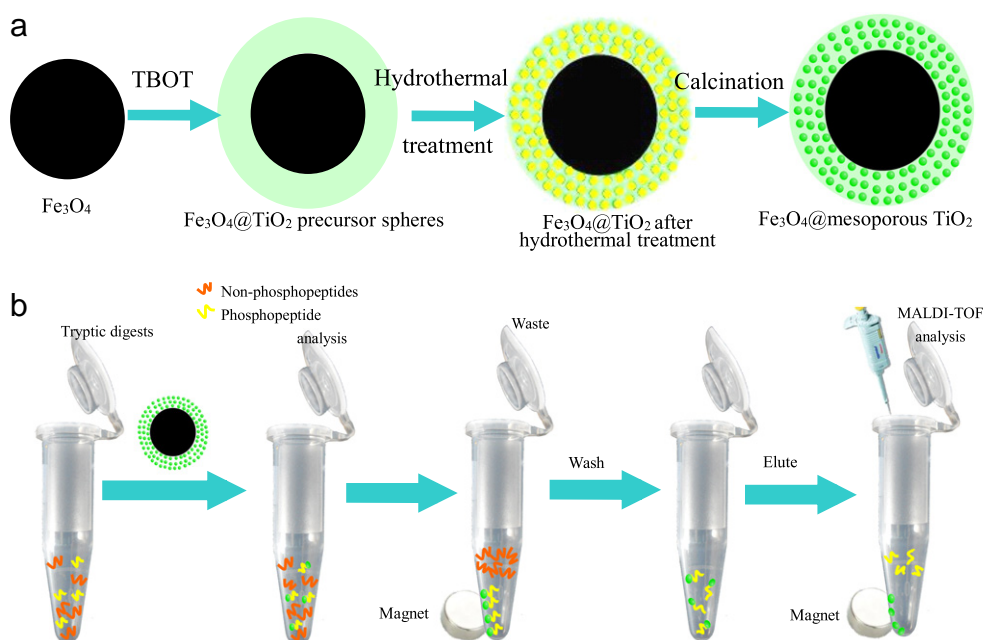
## 2.4. Preparation of the lysate of mouse brain

Mouse brain extract was enriched and analyzed by nano LC-MS/MS. Mice were firstly dissected, and the brain was promptly disemboweled and washed with cold physiological saline to remove the blood and some possible contaminant. The tissue specimens was placed in ice-cold homogenization buffer consisting of 7 M urea, 2 M thiourea and a mixture of protease inhibitor (1 mM phenylmethanesulfonylfluoride) and phosphatase inhibitors (0.2 mM  $\text{Na}_3\text{VO}_4$ , 1 mM NaF). After mincing with scissors and washing to remove blood, the brains were homogenized in a Potter–Elvehjem homogenizer with a Teflon piston, using 5 mL of the homogenization buffer per 1 g of tissue. The suspension was homogenized for approximately 2 min, vortexed at 0 °C for 30 min, and centrifuged at 22,000g for 1.5 h. The supernatant contained the total brain proteins.

Appropriate volumes of protein sample were precipitated as above, lyophilized to dryness, and redissolved in reducing solution (6 M guanidine hydrochloride, 100 mM  $\text{NH}_4\text{HCO}_3$ , pH=8.3) with the protein concentration adjusted to 2  $\mu\text{g}/\mu\text{L}$ . Then, 200  $\mu\text{g}$  of this protein sample (100  $\mu\text{L}$  volume) were mixed with 10  $\mu\text{L}$  of 0.5 M DTT. The mixture was incubated at 37 °C for 1 h, and then 20  $\mu\text{L}$  of 0.5 M 2-iodoacetamide were added and incubated for an additional 30 min at 37 °C in the dark. The protein mixtures were exchanged into 50 mM  $\text{NH}_4\text{HCO}_3$  buffer, pH=8.5, and incubated with trypsin (50:1) at 37 °C overnight.

## 2.5. Selectively enrichment of phosphopeptides from tryptic digestion of standard proteins and mouse brain

For the standard proteins, first, the peptide mixtures originating from tryptic digestions were diluted by 50% acetonitrile and 0.1% TFA aqueous solution (v/v), and suspension of  $\text{Fe}_3\text{O}_4$ @mesoporous  $\text{TiO}_2$  microspheres (400  $\mu\text{g}$ ) was added into 200  $\mu\text{L}$  of diluted peptide mixture. Then the mixed solutions were vibrated at 25 °C for 30 min. After that, with the help of magnet, the pellucid solutions were taken out. After that, the  $\text{Fe}_3\text{O}_4$ @mesoporous  $\text{TiO}_2$  microspheres were washed with 50% acetonitrile and 0.1% TFA aqueous solution (v/v) for three times. Finally the peptides captured by  $\text{Fe}_3\text{O}_4$ @mesoporous  $\text{TiO}_2$  microspheres were



**Scheme 1.** (a) Schematic diagram of the synthetic route of  $\text{Fe}_3\text{O}_4$ @mesoporous  $\text{TiO}_2$ , (b) enrichment protocol for phosphopeptides by using  $\text{Fe}_3\text{O}_4$ @mesoporous  $\text{TiO}_2$ .

eluted with 5  $\mu\text{L}$  of 0.4 M ammonia aqueous solution for 10 min and the eluate was analyzed by MALDI-TOF MS.

The tryptic digests of mouse brain were lyophilized and then dissolved in loading buffer. Two microgram of  $\text{Fe}_3\text{O}_4$ @mesoporous  $\text{TiO}_2$  microspheres were added into 400  $\mu\text{L}$  of diluted mouse brain digests. Then the mixed solutions were vibrated at 25  $^\circ\text{C}$  for 30 min. After that, with the help of magnet, the phosphopeptides adsorbed to  $\text{Fe}_3\text{O}_4$ @mesoporous  $\text{TiO}_2$  microspheres were collected and washed with 50% (v/v) acetonitrile and 0.1% (v/v) TFA aqueous solution for three times. Then the  $\text{Fe}_3\text{O}_4$ @mesoporous  $\text{TiO}_2$  microspheres were eluted with 50  $\mu\text{L}$  of 0.4 M ammonia aqueous solution for 30 min. The eluate was lyophilized and then dissolved in 35  $\mu\text{L}$  loading phase. Finally, the solution was submitted for LC-ESI-MS analysis.

## 2.6. Mass spectrometry and database searching

### 2.6.1. MALDI-TOF MS analysis

For MALDI-TOF MS, sample solutions (0.5  $\mu\text{L}$ ) were deposited on plate using the dried droplet method, and then another 0.5  $\mu\text{L}$  of DHB matrix solution (20 mg/mL, in 50% acetonitrile and 1%  $\text{H}_3\text{PO}_4$  aqueous solution (v/v)) was introduced as a matrix. Analysis of peptides was performed by MALDI-TOF MS on a 5800 Proteomics Analyzer (Applied Biosystems) in the reflector TOF detection modes. The sample was excited using an Nd:YAG laser (383 nm) operated at a repetition rate of 200 Hz and acceleration voltage of 20 kV. All spectra were taken from signal-averaging of 800 laser shots with the laser intensity kept at a proper constant.

### 2.6.2. Nano-LC-ESI-MS/MS analysis

The peptide solution eluted from microspheres were dried thoroughly by lyophilization and then redissolved with 5% ACN aqueous solution containing 0.1% formic acid, separated by nano-LC and analyzed by on-line electrospray tandem mass spectrometry. The experiments were performed on a Nano Aquity UPLC system (Waters Corporation, Milford, USA) connected to an LTQ Orbitrap XL mass spectrometer (Thermo Electron Corp., Bremen, Germany) equipped with an online nanoelectrospray ion source (Michrom Bioresources, Auburn, USA). The separation of the peptides was performed in a Symmetry<sup>®</sup> C18, 5  $\mu\text{m}$ , 180  $\mu\text{m}$  id  $\times$  2 cm trap-column and a BEH300 C18, 1.7  $\mu\text{m}$ , 75  $\mu\text{m}$  id  $\times$  15 cm reverse phase column (Waters Corporation, Milford, USA).

The peptide dissolution were injected onto the trap-column with a flow of 15  $\mu\text{L min}^{-1}$  for 3 min and subsequently eluted with a three-step linear gradient. Starting from 5% B to 45% B in 43 min (mobile phase A: 5% ACN aqueous solution with 0.1%

formic acid; mobile phase B: 95% ACN aqueous solution with 0.1% formic acid), increased to 80% B in 3 min. After holding on 80% B for 1 min, it decreased to the initial condition of 5% B in 1 min. Finally, the column was re-equilibrated at 5% B for 12 min. The column flow rate was maintained at 300 nL/min and column temperature was maintained at 35  $^\circ\text{C}$ . The electrospray voltage of 1.4 kV versus the inlet of the mass spectrometer was used. LTQ Orbitrap XL mass spectrometer was operated in the data-dependent mode to switch automatically between MS and MS/MS acquisition. Survey full-scan MS spectra with two microscans ( $m/z$  400–1800) were acquired in the Orbitrap with a mass resolution of 60,000 at  $m/z$  400, followed by ten sequential LTQ-MS/MS scans. Dynamic exclusion was used with two repeat counts, 10-s repeat duration, and 60-s exclusion duration. For MS/MS, precursor ions were activated using 25% normalized collision energy at the default activation  $q$  of 0.25.

The obtained LTQ Orbitrap XL MS/MS data were identified by using SEQUEST [v.28 (revision 12)], Thermo Electron Corp. against the mouse International Protein Index (IPI) database. To reduce false positive identification results, a decoy database containing the reverse sequences was introduced into the database. The searching parameters were set up as follows: enzyme; the variable modification was phosphorylation (S, T, Y) and oxidation of methionine; the peptide mass tolerance was 100 ppm, and the MS/MS tolerance was 1 Da. And then the obtained searching results were filtrated with following parameters:  $\Delta\text{Cn} > 0.1$ ; the minimum Xcorr of 2.00, and 2.50 corresponding to 2+, and 3+ charge states, respectively. Based on the above parameters, false positive rate of less than 5% was

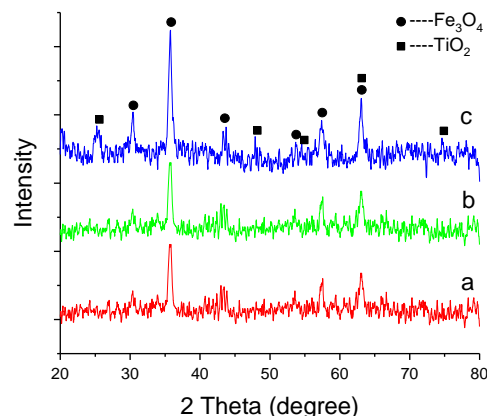


Fig. 2. Wide-angle XRD patterns of (a)  $\text{Fe}_3\text{O}_4$ @ $\text{TiO}_2$  precursor spheres (b)  $\text{Fe}_3\text{O}_4$ @ $\text{TiO}_2$  after hydrothermal treatment (c)  $\text{Fe}_3\text{O}_4$ @mesoporous  $\text{TiO}_2$ .

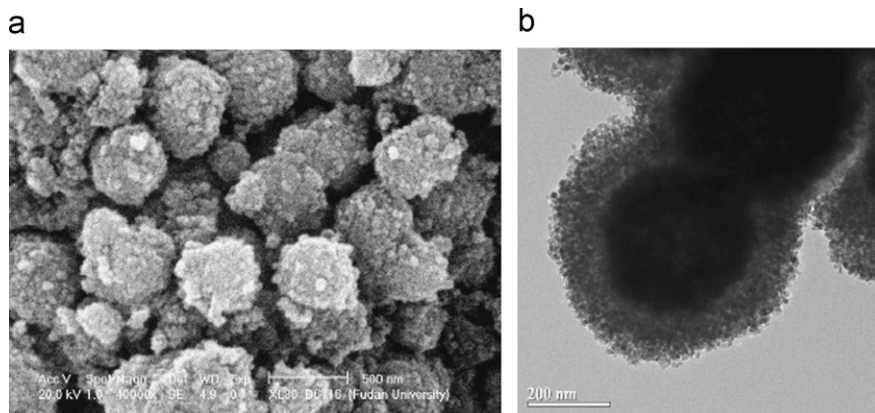
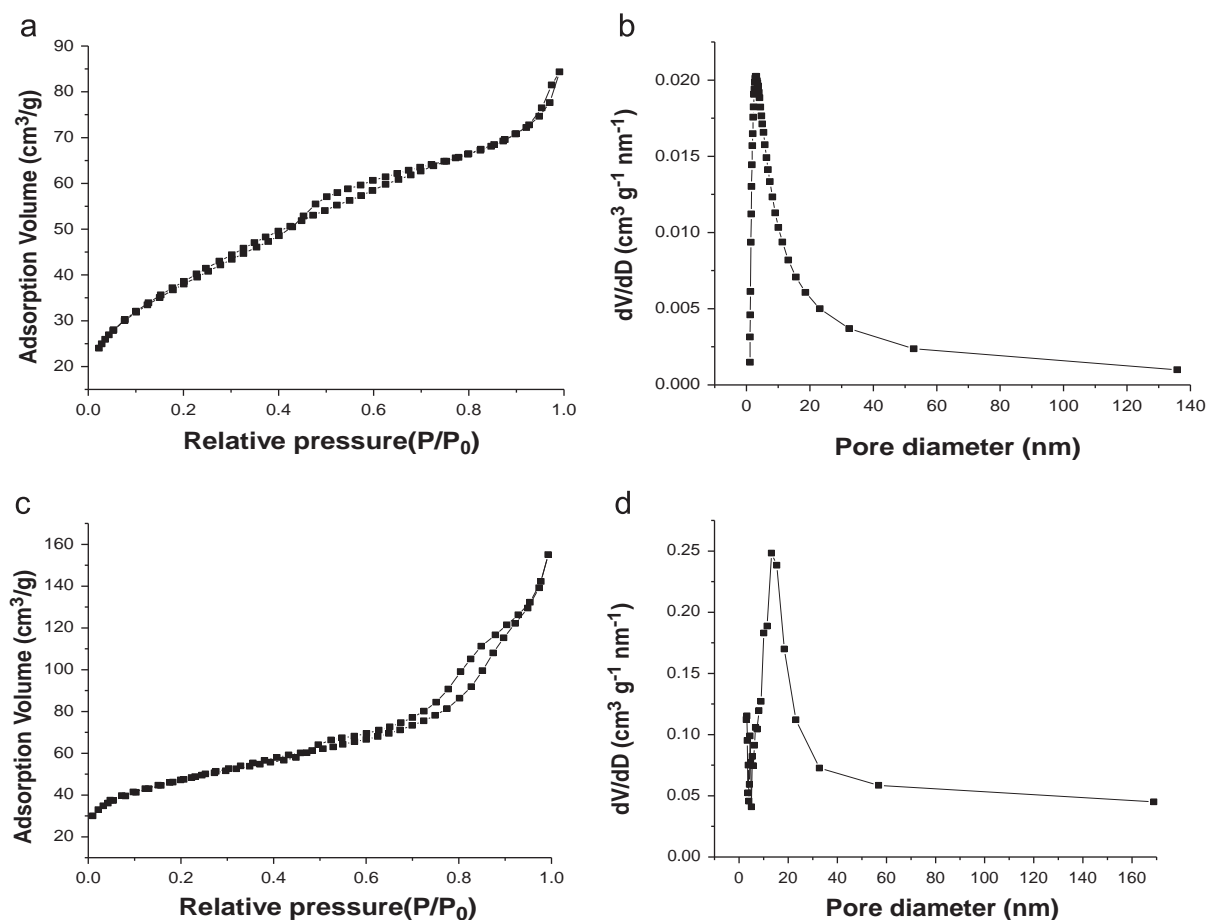
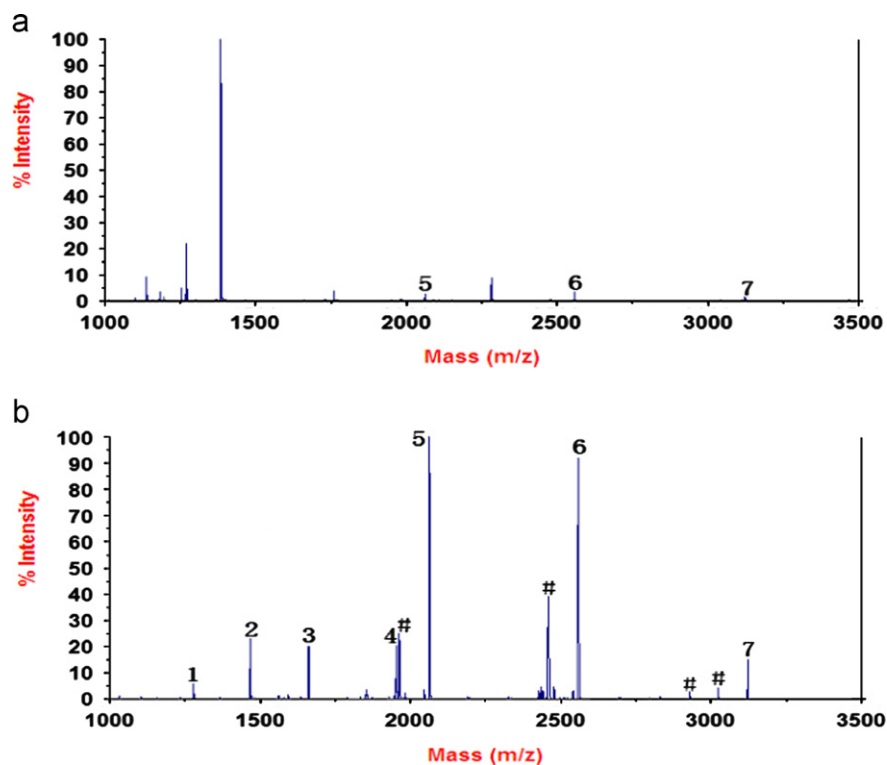


Fig. 1. (a) SEM of  $\text{Fe}_3\text{O}_4$ @mesoporous  $\text{TiO}_2$  (b) TEM of  $\text{Fe}_3\text{O}_4$ @mesoporous  $\text{TiO}_2$ .



**Fig. 3.**  $\text{N}_2$  adsorption–desorption isotherms of (a)  $\text{Fe}_3\text{O}_4@\text{TiO}_2$  precursor spheres (c)  $\text{Fe}_3\text{O}_4@\text{mesoporous TiO}_2$  and pore size distribution of (b)  $\text{Fe}_3\text{O}_4@\text{TiO}_2$  precursor spheres (d)  $\text{Fe}_3\text{O}_4@\text{mesoporous TiO}_2$ .



**Fig. 4.** MALDI-TOF mass spectra of peptides derived from  $\beta$ -casein: (a) without enrichment (b) enriched by  $\text{Fe}_3\text{O}_4@\text{mesoporous TiO}_2$ . Phosphopeptide ions identified are marked with numbers. # the metastable losses of phosphoric acid.

calculated to remove low-probability peptides, demonstrating the reliability of the identified results in this investigation.

### 3. Results and discussion

#### 3.1. Synthesis and characterization of $\text{Fe}_3\text{O}_4$ @mesoporous $\text{TiO}_2$ microspheres

The designed synthesis strategy of  $\text{Fe}_3\text{O}_4$ @mesoporous  $\text{TiO}_2$  microspheres is shown in Scheme 1. First, magnetic  $\text{Fe}_3\text{O}_4$  microspheres were synthesized using our previous reported method, then  $\text{Fe}_3\text{O}_4$ @ $\text{TiO}_2$  precursor spheres were synthesized based on the

**Table 1**

Detail information of the observed phosphopeptides obtained from tryptic digestion of  $\alpha$ -casein S1 and S2, and  $\beta$ -casein<sup>a</sup>.

Peak number	Theoretical $m/z$	Observed $m/z$	aa	Peptide sequence
1	1278.6	1278.9	$\beta/33$ –52	FQSEEQQTDELDQDKIHPF
2	1466.6	1466.7	$\alpha$ -S2/ 153–164	TVDMSTEVFTK
3	1660.8	1660.9	$\alpha$ -S1/106– 119	VPQLEIVPNSAEER
4	1952.0	1952.1	$\alpha$ -S1/104– 119	YKVPQLEIVPNSAEER
5	2061.8	2061.8	$\beta/33$ –48	FQSEEQQTDELDQDK
6	2556.1	2556.2	$\beta/33$ –52	FQSEEQQTDELDQDKIHPF
7	3122.3	3122.3	$\beta/1$ –25	RELEELNVPGEIVESLSSEESITR

<sup>a</sup> The phosphorylation sites are underlined.

hydrolysis and condensation of TBOT. The above obtained  $\text{Fe}_3\text{O}_4$  microspheres (0.15 g) were redispersed in absolute ethanol (200 mL), and mixed with concentrated ammonia solution (0.9 mL, 28 wt%) under ultrasound for 15 min. Afterward, 2.0 mL of TBOT was added dropwise in 5 min, and the reaction was allowed to proceed for 24 h at 45 °C under continuous mechanical stirring. After washing with deionized water and ethanol for 3 times, the above obtained  $\text{Fe}_3\text{O}_4$ @ $\text{TiO}_2$  precursor spheres (0.5 g) were mixed with 20 mL deionized water, then added into a Teflon-lined stainless-steel autoclave (30 mL in capacity). The autoclave was heated at 160 °C for 24 h, and then allowed to cool to room temperature. The resultant product was separated and collected with a magnet, followed by washing with deionized water and ethanol for 3 times, respectively. At last the product was dried under vacuum at 50 °C and calcined in nitrogen at 400 °C for 2 h. After the furnace was left to cool to room temperature, the final products were transferred to tubes and preserved under room temperature.

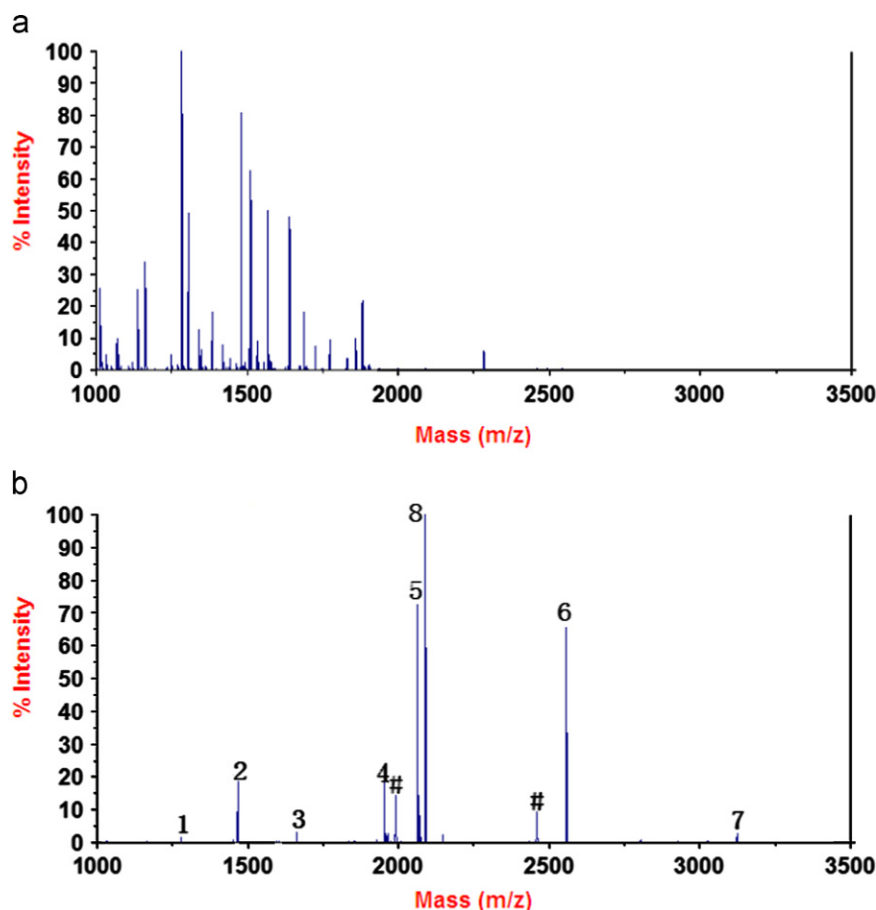
The morphology of prepared microspheres was obtained by scanning electron microscopy (SEM) and transmission electron microscopy (TEM).  $\text{Fe}_3\text{O}_4$ @ $\text{TiO}_2$  precursor spheres are mono-dispersed and have nearly uniform size (Fig. S1a). An obvious

**Table 2**

Detail information of the observed phosphopeptides obtained from tryptic digestion of ovalbumin<sup>a</sup>.

Theoretical $m/z$	Observed $m/z$	aa	Peptide sequence
2088.9	2089.0	Ova/340–359	EVVGSAEAGVDAASVSEEFR

<sup>a</sup> The phosphorylation sites are underlined.



**Fig. 5.** MALDI-TOF mass spectra of peptides derived from a peptide mixture of  $\beta$ -casein, ovalbumin and BSA at a molar ratio of 1:1:50: (a) without enrichment (b) enriched by  $\text{Fe}_3\text{O}_4$ @mesoporous  $\text{TiO}_2$ .



mesoporous  $\text{TiO}_2$  shell ( $\sim 37$  nm) was observed and the mean diameter of the precursor spheres is about 220 nm (Fig. S1b). As shown in Fig. 1a, after solvothermal and calcination treatments of these precursor spheres the monodispersity of the spheres were retained. It is worth nothing that after solvothermal and calcination treatments, amorphous dense  $\text{TiO}_2$  transformed to a large number of crystallization  $\text{TiO}_2$  nanoparticles (Fig. 1b); this was confirmed with the corresponding Wide-angle X-ray diffraction pattern (Fig. 2).

The energy-dispersive X-ray analysis (EDXA) (Fig. S2) of the illuminating electron beams on the obtained  $\text{Fe}_3\text{O}_4$ @mesoporous  $\text{TiO}_2$  reveals the existence of Fe, Ti, Cu and O elements (the Cu peaks in the spectrum originate from the Cu TEM grid), further confirming the successful coating of  $\text{TiO}_2$ .

Wide-angle X-ray diffraction patterns of the  $\text{Fe}_3\text{O}_4$ @ $\text{TiO}_2$  precursor spheres,  $\text{Fe}_3\text{O}_4$ @ $\text{TiO}_2$  after hydrothermal treatment and  $\text{Fe}_3\text{O}_4$ @mesoporous  $\text{TiO}_2$  are shown in Fig. 2. Before calcination,

$\text{TiO}_2$  nanoparticles are amorphous (Fig. 2a and b) and after calcination treatment  $\text{Fe}_3\text{O}_4$ @mesoporous  $\text{TiO}_2$  show well-resolved diffractions peaks which are well-matched to  $\text{Fe}_3\text{O}_4$  (JCPDS 65-3107) and anatase (JCPDS 21-1272) (Fig. 2c), suggesting that their frameworks are crystalline and our obtained products are  $\text{Fe}_3\text{O}_4$ @mesoporous  $\text{TiO}_2$ .

Additionally, the porosity of the microspheres before and after the subsequent solvothermal and calcination treatments was characterized by nitrogen sorption at 77 K. Fig. 3a and b display the  $\text{N}_2$  adsorption–desorption isotherm and the corresponding pore size distribution curve for the  $\text{Fe}_3\text{O}_4$ @ $\text{TiO}_2$  precursor spheres, while Fig. 3c and d display that of  $\text{Fe}_3\text{O}_4$ @mesoporous  $\text{TiO}_2$ . The BET surface area and total pore volume of  $\text{Fe}_3\text{O}_4$ @ $\text{TiO}_2$  precursor spheres were calculated to be  $139.8 \text{ m}^2 \text{ g}^{-1}$  and  $0.13 \text{ cm}^3 \text{ g}^{-1}$ , respectively, and the results of  $\text{Fe}_3\text{O}_4$ @mesoporous  $\text{TiO}_2$  were calculated to be  $162.6 \text{ m}^2 \text{ g}^{-1}$  and  $0.24 \text{ cm}^3 \text{ g}^{-1}$ , respectively. That indicate that the sample has a relatively high surface-to-volume

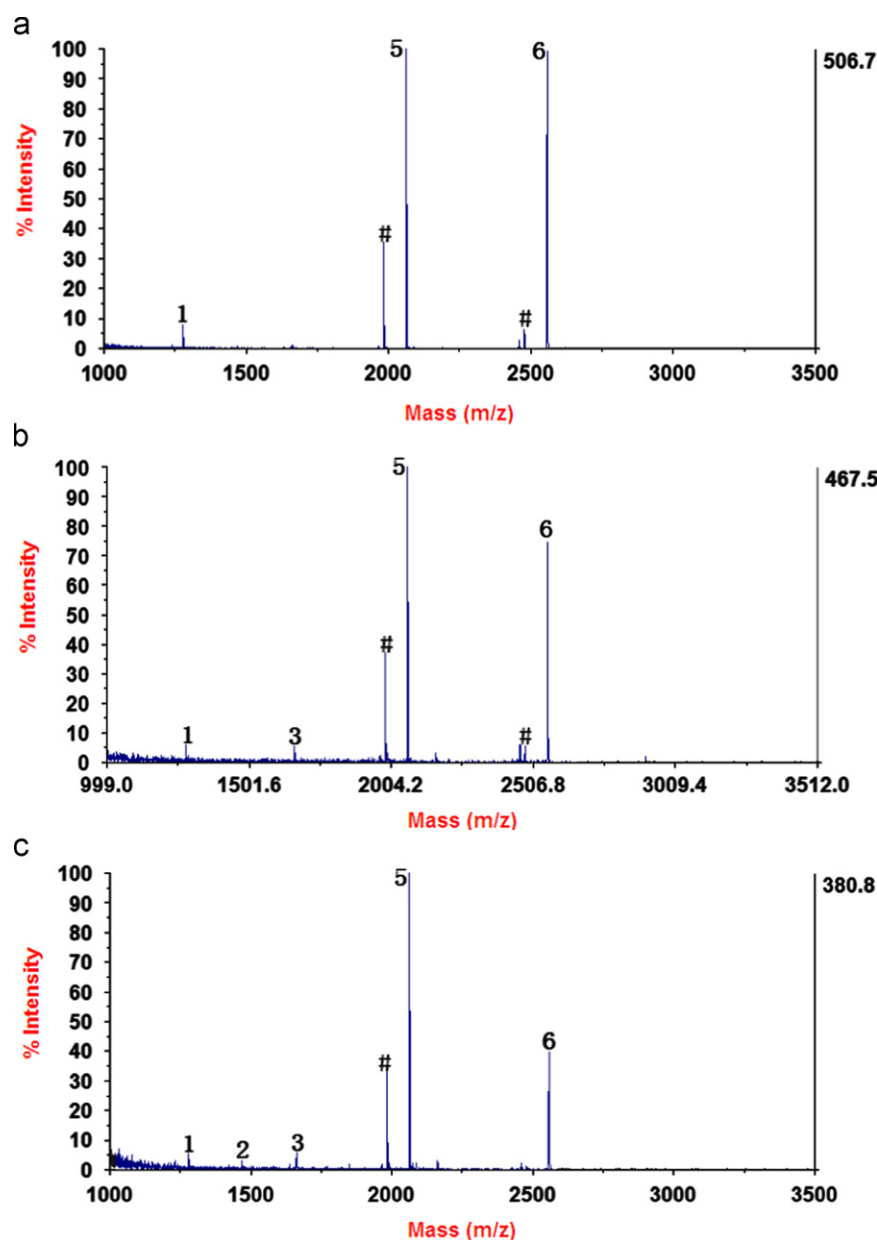


Fig. 6. MALDI-TOF mass spectra of phosphopeptides enriched from  $\beta$ -casein with different concentrations: (a)  $4 \times 10^{-9}$  M (b)  $2 \times 10^{-9}$  M (c)  $4 \times 10^{-10}$  M using  $\text{Fe}_3\text{O}_4$ @mesoporous  $\text{TiO}_2$ . Phosphopeptide ions identified are marked with numbers. # the metastable losses of phosphoric acid.

ratio and after solvothermal and calcination treatments, the specific surface area gets larger, confirming the solvothermal and calcination treatments are necessary to attain a higher surface area [30,31].

### 3.2. Selective enrichment of phosphopeptides from tryptic digestion of standard proteins

The Fe<sub>3</sub>O<sub>4</sub>@mesoporous TiO<sub>2</sub> were first used to isolate phosphopeptides from tryptic digests of  $\beta$ -casein, which is commonly contaminated with a trace of  $\alpha$ -casein. A direct analysis of  $4 \times 10^{-7}$  M  $\beta$ -casein by MALDI-TOF MS indicates that signals of phosphopeptides were feeble and the nonphosphorylated peptides dominate the spectrum (Fig. 4a). However, after enrichment by Fe<sub>3</sub>O<sub>4</sub>@mesoporous TiO<sub>2</sub>, the signals for the  $\beta$ -casein phosphopeptides (marked with numbers) significantly increase and dominate the spectrum (Fig. 4b). Furthermore, three phosphopeptides derived from  $\alpha$ -casein were also observed and nearly no nonphosphorylated peptides were observed, indicating a high enrichment efficiency of the Fe<sub>3</sub>O<sub>4</sub>@mesoporous TiO<sub>2</sub>. The signals marked with a circumflex could be assigned to a dephosphorylated fragment of phosphopeptide through loss of H<sub>3</sub>PO<sub>4</sub>. The observed phosphopeptides are marked with numbers and the detailed information of the observed phosphopeptides is listed in Table 1.

To evaluate the specificity of enrichment of phosphopeptides with Fe<sub>3</sub>O<sub>4</sub>@mesoporous TiO<sub>2</sub>, a relatively complex sample consisting of a peptide mixture originating from a tryptic digest of  $\beta$ -casein ( $4 \times 10^{-8}$  M), ovalbumin ( $4 \times 10^{-8}$  M) and bovine serum albumin ( $2 \times 10^{-6}$  M) with a ratio of 1:1:50, was used as the test sample. The spectrum obtained for MALDI-TOF MS of the peptide mixture is shown in Fig. 5a. The MS spectrum for the analysis of phosphopeptides enriched with Fe<sub>3</sub>O<sub>4</sub>@mesoporous TiO<sub>2</sub> is shown in Fig. 5b. It is obvious that nearly no nonphosphopeptides were described in Fig. 5b and we can see that besides the spectrum of  $\alpha$ -casein and  $\beta$ -casein, we successfully enriched one phosphopeptide of ovalbumin at 2089.0 *m/z*. The detailed information of the observed phosphopeptide of ovalbumin was listed in Table 2. This suggests that Fe<sub>3</sub>O<sub>4</sub>@mesoporous TiO<sub>2</sub> could specifically isolate and enrich the phosphopeptides.

To study the detection limit of the Fe<sub>3</sub>O<sub>4</sub>@mesoporous TiO<sub>2</sub> used for enrichment of phosphopeptides,  $\beta$ -casein digest solutions with different concentrations were applied. The mass spectra are shown in Fig. 6. It can be seen that when the concentration of  $\beta$ -casein digest is  $4 \times 10^{-10}$  M, after enrichment by the Fe<sub>3</sub>O<sub>4</sub>@mesoporous TiO<sub>2</sub>, the ion signals from the phosphopeptides is still powerful, which indicates the high detection sensitivity of the microspheres [32–34]. The results suggest that the Fe<sub>3</sub>O<sub>4</sub>@mesoporous TiO<sub>2</sub> we report here are promising materials for the enrichment of phosphopeptides.

### 3.3. Selective enrichment of phosphopeptides from tryptic digestion of rat brain

To further evaluate the performance of Fe<sub>3</sub>O<sub>4</sub>@mesoporous TiO<sub>2</sub> for phosphopeptide enrichment, the microspheres were employed to enrich phosphopeptides from tryptic digest of mouse brain. After database searching, information of the enriched phosphopeptide was obtained. In total, 1774 phosphorylation sites of 731 phosphopeptides–1082 on serine (60.99%), 510 on threonine (28.75%) and 182 on tyrosine (10.26%)—were identified. The detailed information of the phosphopeptides enriched from tryptic digests of mouse brain by using Fe<sub>3</sub>O<sub>4</sub>@mesoporous TiO<sub>2</sub> are listed in the ESI (see Table S1).

## 4. Conclusions

In summary, we successfully synthesized Fe<sub>3</sub>O<sub>4</sub>@mesoporous TiO<sub>2</sub> with well-defined core shell structure and large specific surface area, and demonstrated their potential for efficient and selective enrichment of phosphopeptides. In the study, Fe<sub>3</sub>O<sub>4</sub>@mesoporous TiO<sub>2</sub> were demonstrated to have the ability for selective enrichment of phosphopeptides by using standard protein digestion. Furthermore, we applied Fe<sub>3</sub>O<sub>4</sub>@mesoporous TiO<sub>2</sub> for selective enrichment of phosphopeptides from tryptic digestion of mouse brain, and the experimental results showed that the approach based on Fe<sub>3</sub>O<sub>4</sub>@mesoporous TiO<sub>2</sub> would be effective for phosphopeptides enrichment.

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## Appendix A. Supporting information

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